

# CBMS208

# **Chemical Analysis I**

S2 Day 2017

Dept of Chemistry & Biomolecular Sciences

# Contents

General Information	2
Learning Outcomes	3
General Assessment Information	3
Assessment Tasks	3
Delivery and Resources	4
Unit Schedule	6
Policies and Procedures	7
Graduate Capabilities	8

#### Disclaimer

Macquarie University has taken all reasonable measures to ensure the information in this publication is accurate and up-to-date. However, the information may change or become out-dated as a result of change in University policies, procedures or rules. The University reserves the right to make changes to any information in this publication without notice. Users of this publication are advised to check the website version of this publication [or the relevant faculty or department] before acting on any information in this publication.

# **General Information**

Unit convenor and teaching staff Lecturer Danny Wong danny.wong@mq.edu.au Contact via danny.wong@mq.edu.au F7B 235

Credit points 3

Prerequisites CBMS101(Cr) or CBMS102

Corequisites

Co-badged status CBMS608

#### Unit description

The measurement of chemical composition is a necessary requirement for forensic science, local and international trade, manufacture and production, government regulatory agencies, biotechnology, and nearly every field of science. Thus, an understanding of the principles of chemical analysis is an essential part of any scientist's education. This unit is designed to serve the needs of students majoring in chemical, biomolecular, environmental, earth and medical sciences. The unit introduces the principles of chemical analysis that enable the separation, detection, identification and quantification of the chemical matter found in a variety of samples. Such samples may range from those associated with forensic science such as drugs in biological tissue to contaminants in river systems, soils and the general environment, to heavy metals in ores and alloys, to neurochemicals present in a single neuron. Topics include: statistical analysis of chemical data; sampling methods; all modes of chromatographic separation with applications to environmental and biotechnological issues; methods of flow analysis of environmental and biological samples; ion-selective electrodes; potentiometric titration; titrimetry and buffer solutions. Understanding of these techniques is reinforced with practical, hands-on experience using instrumentation in our well-equipped analytical teaching laboratory.

#### Important Academic Dates

Information about important academic dates including deadlines for withdrawing from units are available at <a href="https://www.mq.edu.au/study/calendar-of-dates">https://www.mq.edu.au/study/calendar-of-dates</a>

## **Learning Outcomes**

On successful completion of this unit, you will be able to:

- To display competency in executing some commonly used analytical techniques;
- To perform basic analytical experiments with awareness of uncertainty in measurements;
- To interpret and draw sound conclusions from analytical chemical data obtained;
- To prepare written scientific documents at a satisfactory level;
- To develop fundamental interpersonal skills during laboratory sessions.

# **General Assessment Information**

Students who perform well in this unit:

- will have actively participated in ALL laboratory session and completed ALL of the of the assessable items. They will also have taken into account feedback and advice given during the laboratory sessions to improve their performance in these activities throughout the session.
- will have seen improvement in grades throughout session through active engagement. Strong engagement usually results in better understanding of the theory which results in a strong performance in the final exam.

Name	Weighting	Hurdle	Due
Final 3 hour examination	50%	No	University Examination Period
Assignments	15%	No	Fortnightly
Laboratory work	35%	No	2 Weeks after each Lab session

#### Assessment Tasks

#### Final 3 hour examination

# Due: University Examination Period Weighting: 50%

The final examination will be 3 hours in length and will cover all sections of the unit (lectures, tutorial problems, assignments and laboratory exercises).

On successful completion you will be able to:

· To interpret and draw sound conclusions from analytical chemical data obtained;

# Assignments

Due: Fortnightly Weighting: 15%

The assignments are designed to aid in learning the material during the session, rather than trying to cram on the day before the examination. They are relatively low risk (a small component of the aggregate score) but they are very valuable for you as measures of your understanding of the topics. Assignments are performed online.

On successful completion you will be able to:

- To display competency in executing some commonly used analytical techniques;
- To interpret and draw sound conclusions from analytical chemical data obtained;

#### Laboratory work

#### Due: **2 Weeks after each Lab session** Weighting: **35%**

The laboratory sessions provide an opportunity for putting the materials acquired in lectures into practice. They will provide concrete expositions of theory. They also provide the opportunity for developing bench, data collection and data analysis skills. Preparation of written reports based on experimental results obtained will develop some communication skills in the chemistry context.

On successful completion you will be able to:

- · To display competency in executing some commonly used analytical techniques;
- To perform basic analytical experiments with awareness of uncertainty in measurements;
- To interpret and draw sound conclusions from analytical chemical data obtained;
- To prepare written scientific documents at a satisfactory level;
- To develop fundamental interpersonal skills during laboratory sessions.

# **Delivery and Resources**

#### Classes

Timetable: Please check http://www.timetables.mq.edu.au/ for the official timetable of the unit.

Lectures: The material presented in the lectures is important and you should not assume that all examinable material is available in the textbook or in printed notes. On the other hand, do not assume that all examinable material is to be found in the lecture notes.

Tutorial: There are three optional tutorial sessions organised in this unit.

Laboratory Work: A general preparatory workshop for laboratory work will be held in Week 1.

Laboratory sessions commence in Week 2. You will undertake five experiments in the 2nd / 3rd Year Teaching Laboratories.

## **Required and Recommended Texts and/or Materials**

Prescribed text:

- D.C.Harris, Quantitative Chemical Analysis, 9th Edition plus Sapling Access, Macmillan Publishing (2015). ISBN: 9781319044053.
- NOTE: You must purchase the textbook Sapling bundle (available from Co-op Bookshop). The textbook alone is not sufficient as Assignments are distributed through, and require access to, Sapling.

Recommended references (all available in University Library)

- D.A.Skoog, D.M.West, F.J.Holler, S.R.Crouch, Fundamentals of Analytical Chemistry, 9th Edition, Brooks/Cole, Thomson Learning, Inc (2014).
- D.C.Harris, Exploring Chemical Analysis, W.H.Freeman & Company (1997).
- H.H.Willard, L.L.Merritt, Jr., J.A.Dean, F.A.Settle, Jr., Instrumental Methods of Analysis, 7th Edition, Wadworth Publishing Company (1988).
- D.A.Skoog, F.J.Holler and T.A.Nieman, Principles of Instrumental Analysis, 5th Edition, Saunders College Publishing (1998).
- J.F.Rubinson and K.A.Rubinson, Contemporary Chemical Analysis, Prentice Hall (1998).

If you feel you need to strengthen your mathematical skills, you might like to refer to Maths for Chemistry – A Chemist's toolkit of calculations, P.Monk, Oxford University Press (2006).

A useful guidebook for studying science subjects is Essential Skills for Science and Technology, P.Zeegers, K.Deller-Evans, S.Egege, C.Klinger, Oxford University Press (2008).

## **Technology Used and Required**

The web page for this unit can be found at: http://ilearn.mq.edu.au

Lecture notes will be available on the Web for downloading one week prior to the scheduled lecture. You are strongly encouraged to make use the discussion forum available on the CBMS208 website for general discussion of materials presented in this unit.

# **Teaching and Learning Strategy**

Students are required to attend lectures and laboratory classes. Active participation by the students in all of these fora is expected. This means that you are expected to ask questions during lectures, and particularly in laboratory sessions. Learning is an active process, and as such, you must engage with the material. This means reading the textbook (and beyond) before and after lectures, attempt the assignment questions and other questions, discuss the concepts with your classmates and lecturers. Do not be afraid to ask questions – your classmates will probably want to ask the same thing.

Assignment questions are issued so that you will have the opportunity to use the information provided in the lectures and textbook and to test your degree of understanding of those topics.

Laboratory exercises are designed to provide a concrete example of the abstract topics covered in the course work, and to give you the opportunity to discover the principles and applications for yourself. Laboratory exercises also offer the opportunity to explore the uncertainty inherent in scientific investigations and the limitations of models and theories by allowing comparison with real systems.

# **Unit Schedule**

Note: This schedule in tenative and may be subject to change

Week	Week starting	Monday 9 am	Monday 11 am	Thursday 4 pm	LABORATORY 9am – 1pm/ 2 – 6 pm E7B 349
1	July 31	Introduction, Statistics	Statistics	Statistics	Workshop (E7B 346 or F7B 322)
2	August 7	Statistics	Statistics	Statistics	Experiment 1 (E7B 349)
3	August 14	Statistics	Sampling methods	Sampling methods	Experiment 2 (E7B 349)
4	August 21	Titrimetry	Titrimetry	Titrimetry	-
5	August 28	Titrimetry	Titrimetry	Titrimetry	Experiment 3 (E7B 349)
6	September 4	Flow injection	Flow injection	Flow injection	Optional Tutorial
7	September 11	Flow injection	Electroanalytical chemistry	Electroanalytical chemistry	Experiment 4(i) / 4(ii) / 5
	Mid-Session Break (September 18 – September 29)				
8	October 2	No Lecture Labour Day holiday	No Lecture Labour Day holiday	Electroanalytical chemistry	Experiment 4(i) / 4(ii) / 5
9	October 9	Electroanalytical chemistry	Electroanalytical chemistry	Electroanalytical chemistry	Optional Tutorial
10	October 16	Electroanalytical chemistry	Chromatography	Chromatography	Experiment 4(i) / 4(ii) / 5

11	October 23	Chromatography	Chromatography	Chromatography	_
12	October 30	Chromatography	Chromatography	Chromatography	Experiment 4(i) / 4(ii) / 5
13	November 6	Chromatography	Chromatography	Optional Tutorial	Optional Tutorial

# **Policies and Procedures**

Macquarie University policies and procedures are accessible from <u>Policy Central</u>. Students should be aware of the following policies in particular with regard to Learning and Teaching:

Academic Honesty Policy http://mq.edu.au/policy/docs/academic\_honesty/policy.html

Assessment Policy http://mq.edu.au/policy/docs/assessment/policy\_2016.html

Grade Appeal Policy http://mq.edu.au/policy/docs/gradeappeal/policy.html

Complaint Management Procedure for Students and Members of the Public <u>http://www.mq.edu.a</u> u/policy/docs/complaint\_management/procedure.html

Disruption to Studies Policy (in effect until Dec 4th, 2017): <u>http://www.mq.edu.au/policy/docs/disr</u>uption\_studies/policy.html

Special Consideration Policy (in effect from Dec 4th, 2017): <u>https://staff.mq.edu.au/work/strategy-</u>planning-and-governance/university-policies-and-procedures/policies/special-consideration

In addition, a number of other policies can be found in the <u>Learning and Teaching Category</u> of Policy Central.

#### **Student Code of Conduct**

Macquarie University students have a responsibility to be familiar with the Student Code of Conduct: https://students.mq.edu.au/support/student\_conduct/

#### Results

Results shown in *iLearn*, or released directly by your Unit Convenor, are not confirmed as they are subject to final approval by the University. Once approved, final results will be sent to your student email address and will be made available in <u>eStudent</u>. For more information visit <u>ask.m</u> <u>q.edu.au</u>.

#### Student Support

Macquarie University provides a range of support services for students. For details, visit <u>http://stu</u> dents.mq.edu.au/support/

#### Learning Skills

Learning Skills (mq.edu.au/learningskills) provides academic writing resources and study strategies to improve your marks and take control of your study.

Workshops

- StudyWise
- Academic Integrity Module for Students
- Ask a Learning Adviser

## Student Services and Support

Students with a disability are encouraged to contact the **Disability Service** who can provide appropriate help with any issues that arise during their studies.

#### **Student Enquiries**

For all student enquiries, visit Student Connect at ask.mq.edu.au

# IT Help

For help with University computer systems and technology, visit <u>http://www.mq.edu.au/about\_us/</u>offices\_and\_units/information\_technology/help/.

When using the University's IT, you must adhere to the <u>Acceptable Use of IT Resources Policy</u>. The policy applies to all who connect to the MQ network including students.

# **Graduate Capabilities**

#### Discipline Specific Knowledge and Skills

Our graduates will take with them the intellectual development, depth and breadth of knowledge, scholarly understanding, and specific subject content in their chosen fields to make them competent and confident in their subject or profession. They will be able to demonstrate, where relevant, professional technical competence and meet professional standards. They will be able to articulate the structure of knowledge of their discipline, be able to adapt discipline-specific knowledge to novel situations, and be able to contribute from their discipline to inter-disciplinary solutions to problems.

This graduate capability is supported by:

#### Learning outcomes

- To display competency in executing some commonly used analytical techniques;
- To perform basic analytical experiments with awareness of uncertainty in measurements;
- To prepare written scientific documents at a satisfactory level;

#### Assessment tasks

- Final 3 hour examination
- Assignments
- Laboratory work

# Critical, Analytical and Integrative Thinking

We want our graduates to be capable of reasoning, questioning and analysing, and to integrate

and synthesise learning and knowledge from a range of sources and environments; to be able to critique constraints, assumptions and limitations; to be able to think independently and systemically in relation to scholarly activity, in the workplace, and in the world. We want them to have a level of scientific and information technology literacy.

This graduate capability is supported by:

#### Learning outcomes

- To perform basic analytical experiments with awareness of uncertainty in measurements;
- To interpret and draw sound conclusions from analytical chemical data obtained;
- To prepare written scientific documents at a satisfactory level;

#### Assessment tasks

- Final 3 hour examination
- Assignments
- Laboratory work

# Problem Solving and Research Capability

Our graduates should be capable of researching; of analysing, and interpreting and assessing data and information in various forms; of drawing connections across fields of knowledge; and they should be able to relate their knowledge to complex situations at work or in the world, in order to diagnose and solve problems. We want them to have the confidence to take the initiative in doing so, within an awareness of their own limitations.

This graduate capability is supported by:

#### Learning outcomes

- To interpret and draw sound conclusions from analytical chemical data obtained;
- To prepare written scientific documents at a satisfactory level;

#### **Assessment tasks**

- Assignments
- Laboratory work

## Effective Communication

We want to develop in our students the ability to communicate and convey their views in forms effective with different audiences. We want our graduates to take with them the capability to read, listen, question, gather and evaluate information resources in a variety of formats, assess, write clearly, speak effectively, and to use visual communication and communication technologies as appropriate.

This graduate capability is supported by:

#### Learning outcomes

- To prepare written scientific documents at a satisfactory level;
- To develop fundamental interpersonal skills during laboratory sessions.

#### **Assessment task**

• Laboratory work

#### Socially and Environmentally Active and Responsible

We want our graduates to be aware of and have respect for self and others; to be able to work with others as a leader and a team player; to have a sense of connectedness with others and country; and to have a sense of mutual obligation. Our graduates should be informed and active participants in moving society towards sustainability.

This graduate capability is supported by:

#### Learning outcome

• To develop fundamental interpersonal skills during laboratory sessions.