MECH3001
Thermodynamics
Session 1, In person-scheduled-weekday, North Ryde 2024
School of Engineering

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Macquarie University has taken all reasonable measures to ensure the information in this publication is accurate and up-to-date. However, the information may change or become out-dated as a result of change in University policies, procedures or rules. The University reserves the right to make changes to any information in this publication without notice. Users of this publication are advised to check the website version of this publication [or the relevant faculty or department] before acting on any information in this publication.
General Information

Unit convenor and teaching staff
Nazmul Huda
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Contact via +61 2 9850 2249
50 Waterloo Road, Room 131
Thursday 2.00 - 4.00

Credit points
10

Prerequisites
(MECH2002 or MECH202 and (20cp at 2000 level or above)) or Admission to MEngMechEng

Corequisites

Co-badged status

Unit description
This unit examines the principles of thermodynamics. The unit covers knowledge in energy, enthalpy, entropy, energy transfer, mass and energy balance, laws of thermodynamics, and the design principles of thermo-fluid systems. At the end of the unit, students are expected to demonstrate the ability to improve the design of real-world thermo-fluid systems and demonstrate a detailed understanding and the application of energy systems.

Important Academic Dates
Information about important academic dates including deadlines for withdrawing from units are available at https://www.mq.edu.au/study/calendar-of-dates

Learning Outcomes
On successful completion of this unit, you will be able to:

ULO1: Articulate and interpret the Laws of Thermodynamics and Energy systems including properties of substances, state and equilibrium
ULO2: Apply the concept of energy, enthalpy, entropy, entropy balance, energy balance and energy transfer in analysing thermodynamic systems
ULO3: Analyse mass and energy transfer in both closed and open systems in steady and unsteady states.
ULO4: Examine gas power cycle, vapour and combined power cycle, refrigeration cycle
and air-conditioning cycle and apply knowledge of thermodynamics to improve and optimise the design of the existing cycle.

ULO5: Exhibit specific skills in teamwork and written communication skills through technical report writing and laboratory work.

General Assessment Information

Student Responsibilities

Be familiar with University policy and College procedures and act in accordance with those policies and procedures.

It is the responsibility of the student to retain a copy of any work submitted. Students must produce these documents upon request. Copies should be retained until the end of the grade appeal period each term.

The student is to perform the required due diligence for their assessment grade and rectify as soon as possible upon finding any errors.

Late Assessment Submission Penalty

Students enrolled in Session-based units with written assessments will have the following university standard late penalty applied. Please see https://students.mq.edu.au/study/assessments for more information.

Unless a Special Consideration request has been submitted and approved, a 5% penalty (of the total possible mark) will be applied each day a written assessment is not submitted, up until the 7th day (including weekends). After the 7th day, a grade of '0' will be awarded even if the assessment is submitted. Submission time for all written assessments is set at 11:55 pm. A 1-hour grace period is provided to students who experience a technical concern.

For any late submission of time-sensitive tasks, such as scheduled tests/exams, performance assessments/presentations, and/or scheduled practical assessments/labs, students need to submit an application for Special Consideration.

Practical Component

Attendance of practical classes is mandatory before submitting the lab report. Lab reports submitted without attending the practical session will get a grade of '0' even if the assessment is submitted by the due date. A special consideration request must be submitted and approved if any student is unable to attend a practical session to organize alternative arrangements.

Resubmission option
Resubmission of any assessment task is not allowed under any circumstances.

Notifications

Formal notification of assessment tasks, grading rubrics, and due dates will be posted on iLearn. Although all reasonable measures to ensure the information is accurate, The University reserves the right to make changes without notice. Each student is responsible for checking iLearn for changes and updates.

Report and Assignment Tasks

Assignment Problems will be posted on iLearn at least one week before their submission date. Assignment solutions will be posted within a week after the submission date. Submissions will not be accepted once the solution is posted.

Assignment submissions and plagiarism policies

All assignments and reports must be submitted electronically through iLearn (in pdf format). Submissions will undergo plagiarism checkers using the Turnitin software and any work deemed to have a 30% or higher similarity score may incur an academic penalty. For more details on the policies of academic penalties relating to academic honesty, please refer to the policies and procedures section below.

Submissions are expected to be typed set in a logical layout and sequence. Markers WILL NOT grade poorly organized or illegible scans or drafts. The expected workload includes the preparation of final copies and clear diagrams.

Grading and passing requirements for unit

For further details about grading, please refer below to the policies and procedures section.

In order to pass this unit, a student must obtain a mark of 50 or more for the unit (i.e. obtain a passing grade P/ CR/ D/ HD).

The unit will be graded according to the Macquarie University Grading policy. The following grades will be used according to the listed numerical range:

ASSESSMENT GRADES AND STATUS

<table>
<thead>
<tr>
<th>GRADE</th>
<th>RANGE</th>
<th>STATUS (‘Standard Grade’ in AMIS)</th>
<th>DESCRIPTION</th>
</tr>
</thead>
</table>

Assessment Tasks

<table>
<thead>
<tr>
<th>Name</th>
<th>Weighting</th>
<th>Hurdle</th>
<th>Due</th>
</tr>
</thead>
<tbody>
<tr>
<td>Laboratory report</td>
<td>18%</td>
<td>No</td>
<td>Week 6, Week 9, Week 12</td>
</tr>
<tr>
<td>In class midterm test during lecture</td>
<td>10%</td>
<td>No</td>
<td>Week 7</td>
</tr>
</tbody>
</table>

Final Examinations

Final examinations will typically take place at the end of the semester. For further information, please refer to the Examination Timetable website on [www.mq.edu.au](http://www.mq.edu.au).

If you receive special consideration for the final exam, a supplementary exam will be scheduled by the faculty during a supplementary exam period, typically about 3 to 4 weeks after the normal exam period. By making a special consideration application for the final exam you are declaring yourself available for a resit during the supplementary examination period and will not be eligible for a second special consideration approval based on pre-existing commitments. Please ensure you are familiar with the policy prior to submitting an application. Approved applicants will receive an individual notification one week prior to the exam with the exact date and time of their supplementary examination.

Assessment Tasks
### Laboratory report
Assessment Type 1: Lab report
Indicative Time on Task 2: 10 hours
Due: **Week 6, Week 9, Week 12**
Weighting: **18%**

Laboratory report based on the practical classes

On successful completion you will be able to:
- Exhibit specific skills in teamwork and written communication skills through technical report writing and laboratory work.

### In class midterm test during lecture
Assessment Type 1: Quiz/Test
Indicative Time on Task 2: 10 hours
Due: **Week 7**
Weighting: **10%**

Midterm in class examination

On successful completion you will be able to:
- Articulate and interpret the Laws of Thermodynamics and Energy systems including properties of substances, state and equilibrium
- Apply the concept of energy, enthalpy, entropy, entropy balance, energy balance and energy transfer in analysing thermodynamic systems
- Analyse mass and energy transfer in both closed and open systems in steady and
unsteady states.

- Examine gas power cycle, vapour and combined power cycle, refrigeration cycle and air-conditioning cycle and apply knowledge of thermodynamics to improve and optimise the design of the existing cycle.

**Weekly SGTA classes**

Assessment Type 1: Participatory task  
Indicative Time on Task 2: 10 hours  
Due: **Week 2 to Week 13**  
Weighting: 6%

Development of knowledge and skills requires continual practice at authentic tasks. In weekly SGTA classes, you will undertake a range of activities and complete various worksheets.

On successful completion you will be able to:

- Articulate and interpret the Laws of Thermodynamics and Energy systems including properties of substances, state and equilibrium
- Apply the concept of energy, enthalpy, entropy, entropy balance, energy balance and energy transfer in analysing thermodynamic systems
- Analyse mass and energy transfer in both closed and open systems in steady and unsteady states.
- Examine gas power cycle, vapour and combined power cycle, refrigeration cycle and air-conditioning cycle and apply knowledge of thermodynamics to improve and optimise the design of the existing cycle.
- Exhibit specific skills in teamwork and written communication skills through technical report writing and laboratory work.

**Assignment based on problem solving**

Assessment Type 1: Problem set  
Indicative Time on Task 2: 5 hours  
Due: **Week 8 and Week 13**  
Weighting: **16%**

Assignment based on problem solving activities
On successful completion you will be able to:

• Articulate and interpret the Laws of Thermodynamics and Energy systems including properties of substances, state and equilibrium
• Apply the concept of energy, enthalpy, entropy, entropy balance, energy balance and energy transfer in analysing thermodynamic systems
• Analyse mass and energy transfer in both closed and open systems in steady and unsteady states.
• Examine gas power cycle, vapour and combined power cycle, refrigeration cycle and air-conditioning cycle and apply knowledge of thermodynamics to improve and optimise the design of the existing cycle.

Final Examination

Assessment Type 1: Examination
Indicative Time on Task 2: 30 hours
Due: During final examination period
Weighting: 50%

On successful completion you will be able to:

• Articulate and interpret the Laws of Thermodynamics and Energy systems including properties of substances, state and equilibrium
• Apply the concept of energy, enthalpy, entropy, entropy balance, energy balance and energy transfer in analysing thermodynamic systems
• Analyse mass and energy transfer in both closed and open systems in steady and unsteady states.
• Examine gas power cycle, vapour and combined power cycle, refrigeration cycle and air-conditioning cycle and apply knowledge of thermodynamics to improve and optimise the design of the existing cycle.

1 If you need help with your assignment, please contact:

• the academic teaching staff in your unit for guidance in understanding or completing this type of assessment
• the Writing Centre for academic skills support.
Indicative time-on-task is an estimate of the time required for completion of the assessment task and is subject to individual variation.

**Delivery and Resources**

**Primary Text:** Thermodynamics: An Engineering Approach (9th Edition in SI Units) by Yunus A. Cengel and Michael A. Boles

**Supporting Texts:** 1. Engineering Thermodynamics (4th Edition) by Rogers and Mayhew
2. Principles of Engineering Thermodynamics (7th Edition) by Moran, Shapiro, Boettner and Bailey

**Unit Schedule**

<table>
<thead>
<tr>
<th>Week</th>
<th>Lecture Topics</th>
<th>Key Topics to be Covered</th>
</tr>
</thead>
<tbody>
<tr>
<td>1</td>
<td>Introduction and Basic Concepts</td>
<td>Thermodynamics and Energy, Dimension and Units, Different Applications and definitions related to Thermodynamics, Processes and Cycle</td>
</tr>
<tr>
<td>2</td>
<td>Energy, Energy Transfer and General Energy Analysis</td>
<td>Forms of energy, Energy transfer by work and heat, First law of Thermodynamics, Energy efficiency</td>
</tr>
<tr>
<td>3</td>
<td>Properties of pure substances</td>
<td>Properties of pure substances, Phase change processes, Property diagram and property tables, Equation of state</td>
</tr>
<tr>
<td>4</td>
<td>Energy Analysis of Closed Systems</td>
<td>Closed system, Moving boundary work, Energy balance for Closed systems, Internal energy, enthalpy and specific heats</td>
</tr>
<tr>
<td>7</td>
<td>Entropy</td>
<td>Entropy, Entropy diagrams, Entropy change, Entropy balance,</td>
</tr>
</tbody>
</table>
Macquarie University policies and procedures are accessible from Policy Central (https://policies.mq.edu.au). Students should be aware of the following policies in particular with regard to Learning and Teaching:

- Academic Appeals Policy
- Academic Integrity Policy
- Academic Progression Policy
- Assessment Policy
- Fitness to Practice Procedure
- Assessment Procedure
- Complaints Resolution Procedure for Students and Members of the Public
- Special Consideration Policy

Students seeking more policy resources can visit Student Policies (https://students.mq.edu.au/support/study/policies). It is your one-stop-shop for the key policies you need to know about throughout your undergraduate student journey.

To find other policies relating to Teaching and Learning, visit Policy Central (https://policies.mq.edu.au) and use the search tool.
**Student Code of Conduct**

Macquarie University students have a responsibility to be familiar with the Student Code of Conduct: [https://students.mq.edu.au/admin/other-resources/student-conduct](https://students.mq.edu.au/admin/other-resources/student-conduct)

**Results**

Results published on platform other than eStudent, (eg. iLearn, Coursera etc.) or released directly by your Unit Convenor, are not confirmed as they are subject to final approval by the University. Once approved, final results will be sent to your student email address and will be made available in eStudent. For more information visit [ask.mq.edu.au](http://ask.mq.edu.au) or if you are a Global MBA student contact [globalmba.support@mq.edu.au](mailto:globalmba.support@mq.edu.au)

**Academic Integrity**

At Macquarie, we believe academic integrity – honesty, respect, trust, responsibility, fairness and courage – is at the core of learning, teaching and research. We recognise that meeting the expectations required to complete your assessments can be challenging. So, we offer you a range of resources and services to help you reach your potential, including free online writing and maths support, academic skills development and wellbeing consultations.

**Student Support**

Macquarie University provides a range of support services for students. For details, visit [http://students.mq.edu.au/support/](http://students.mq.edu.au/support/)

**The Writing Centre**

The Writing Centre provides resources to develop your English language proficiency, academic writing, and communication skills.

- **Workshops**
- **Chat with a WriteWISE peer writing leader**
- **Access StudyWISE**
- **Upload an assignment to Studiosity**
- **Complete the Academic Integrity Module**

The Library provides online and face to face support to help you find and use relevant information resources.

- **Subject and Research Guides**
- **Ask a Librarian**

**Student Services and Support**

Macquarie University offers a range of Student Support Services including:

- **IT Support**
- **Accessibility and disability support with study**
**Student Enquiries**

Got a question? Ask us via [AskMQ](http://www.mq.edu.au/about_us/offices_and_units/information_technology/help/), or contact [Service Connect](http://www.mq.edu.au/about_us/offices_and_units/information_technology/help/).

**IT Help**

For help with University computer systems and technology, visit [http://www.mq.edu.au/about_us/offices_and_units/information_technology/help/](http://www.mq.edu.au/about_us/offices_and_units/information_technology/help/).

When using the University's IT, you must adhere to the [Acceptable Use of IT Resources Policy](http://www.mq.edu.au/about_us/offices_and_units/information_technology/help/). The policy applies to all who connect to the MQ network including students.

**Engineers Australia Competency Mapping**

<table>
<thead>
<tr>
<th>EA Competency Standard</th>
<th>Unit Learning Outcomes</th>
</tr>
</thead>
<tbody>
<tr>
<td><strong>Knowledge and Skill Base</strong></td>
<td></td>
</tr>
<tr>
<td>1.1 Comprehensive, theory-based understanding of the underpinning fundamentals applicable to the engineering discipline.</td>
<td>1, 2, 3, 4, 5</td>
</tr>
<tr>
<td>1.2 Conceptual understanding of underpinning maths, analysis, statistics, computing.</td>
<td>1, 2, 3, 4, 5</td>
</tr>
<tr>
<td>1.3 In-depth understanding of specialist bodies of knowledge</td>
<td>1, 2, 3, 4, 5</td>
</tr>
<tr>
<td>1.4 Discernment of knowledge development and research directions</td>
<td></td>
</tr>
<tr>
<td>1.5 Knowledge of engineering design practice</td>
<td>1, 2, 3, 4</td>
</tr>
<tr>
<td>1.6 Understanding of scope, principles, norms, accountabilities of sustainable engineering practice.</td>
<td></td>
</tr>
<tr>
<td><strong>Engineering Application Ability</strong></td>
<td></td>
</tr>
<tr>
<td>2.1 Application of established engineering methods to complex problem solving</td>
<td>1, 2, 3, 4, 5</td>
</tr>
<tr>
<td>2.2 Fluent application of engineering techniques, tools and resources.</td>
<td>1, 2, 3, 4, 5</td>
</tr>
<tr>
<td>2.3 Application of systematic engineering synthesis and design processes.</td>
<td>1, 2, 3, 4, 5</td>
</tr>
<tr>
<td>2.4 Application of systematic approaches to the conduct and management of engineering projects.</td>
<td></td>
</tr>
</tbody>
</table>

[https://unitguides.mq.edu.au/unit_offerings/164160/unit_guide/print](https://unitguides.mq.edu.au/unit_offerings/164160/unit_guide/print)
<table>
<thead>
<tr>
<th>Professional and Personal Attributes</th>
<th>3.1 Ethical conduct and professional accountability.</th>
<th>5</th>
</tr>
</thead>
<tbody>
<tr>
<td>3.2 Effective oral and written communication in professional and lay domains.</td>
<td>5</td>
<td></td>
</tr>
<tr>
<td>3.3 Creative, innovative and pro-active demeanour.</td>
<td></td>
<td></td>
</tr>
<tr>
<td>3.4 Professional use and management of information.</td>
<td>5</td>
<td></td>
</tr>
<tr>
<td>3.5 Orderly management of self, and professional conduct.</td>
<td>5</td>
<td></td>
</tr>
<tr>
<td>3.6 Effective team membership and team leadership</td>
<td>5</td>
<td></td>
</tr>
</tbody>
</table>

Unit information based on version 2024.01R of the Handbook